

A Novel Method for Obtaining Carbon Molecular Sieves for N₂/O₂ Separation

Koki Urita¹, Kaito Marubayashi¹, Miyu Hamasaki¹, Hideki Tanaka², Takashi Ishida³, Yasuyuki Yamane³, Jin Miyawaki⁴, Hiroo Notohara¹, Isamu Moriguchi¹

¹ Department of Engineering, Nagasaki University, 1-14 Bunkyo-machi, Nagasaki 852-8521, Japan

² Research Initiative for Supra-Materials (RISM), Shinshu University, 4-17-1 Wakasato, Nagano 380-8553, Japan

³ Research & Development Dept., Activated Carbon Business Div., Osaka Gas Chemicals Co., Ltd., 5-11-61, Torishima, Konohana-ku, Osaka, 554-0051, Japan

⁴ Institute for Materials Chemistry and Engineering, Kyushu University, 6-1 KasugaKoen, Kasuga, Fukuoka 816-8580, Japan

email for correspondence: urita@nagasaki-u.ac.jp

The use of oxygen-enriched air with high oxygen concentration is expected to reduce fuel consumption in industrial furnaces. The pressure swing adsorption method was used for the air separation. However, temperature swing adsorption (TSA) using combustion exhaust heat is expected to save more energy. To achieve highly efficient air separation by introducing TSA, a molecular sieve adsorbent whose pore entrance diameter is optimized for preferential adsorption of oxygen is required [1]. Because the electrochemical oxidation (ECO) process adds oxygen-containing functional groups (OCFGs) to carbon materials [2], ECO can be expected to narrow the pore entrance. ECO is usually performed using block carbon electrodes obtained by mixing with a binder. On the other hand, a flow-type ECO is operated by pumping the suspension into the electric field using a suspension of the electrolyte and carbon materials. In this study, we examined the conditions for the addition of OCFG to porous carbons by flow-type ECO and aimed to create a molecular sieving carbon with oxygen/nitrogen selectivity. XPS analysis indicated that the oxygen content on the carbon surface was increased by ECO, and the addition of OCFGs to the porous carbon surface is possible in the flow-type ECO method even though the carbon electrode is not fixed on a current collector. O₂ adsorption isotherms at 20 °C and 80 °C showed that effective O₂ adsorption was decreased in the ECO-treated samples. The decrease in the adsorbed amount should stem from partial pore occlusion and pore size reduction by the addition of OCFGs to the pores. However, by comparing the adsorption time required to reach the adsorbed amount to half the equilibrium adsorption amount before and after ECO treatment, it was found that the adsorption rate of N₂ was significantly reduced by the ECO treatment. These results suggest that the addition of OCFGs to the pore entrance of porous carbon by ECO is effective for the fabrication of carbon molecular sieves for oxygen/nitrogen separation.

References:

1. Y. Yamane, M. Miyahara, H. Tanaka, *ACS Applied Material Interfaces*, 14 (2022) 177878-17888.
2. C.A. Leon and L.R. Radovic, *Interfacial Chemistry and Electrochemistry of Carbon Surfaces*. In: P.A. Throver (Ed.), *Chemistry and Physics of Carbon*, 24, New York; Marcel Dekker; 1994 p. 213-310.